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NH₃ has 4 negative centres of charge/three bonding pairs and one lone pair and BF₃ has 3 negative centres of

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charge/three bonding pairs / OWTTE; (bond angles) 107° in NH_3 and 120° in BF_3 ; [4] Accept 107.5° for NH_3 . (iv) BF_3 not polar as no net dipole moment / BF bond polarities cancel each other out / symmetrical distribution of charge;

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