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 $\tau = 1/\lambda$ (b) Let h be the
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a quantity to decay to one-
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1.4 State: Montana Year Pop (thousands)

1870	21	1880	39
1890	143	1900	243
1910	376	1920	549
1930	538	1940	559
1950	591	1960	675
1970	694	1980	787
1990	799	2000	902

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Let h be the half-life, that is, the amount of time it takes for a quantity to decay to one-half of its original amount. Since $\lambda = 1/\tau$, we get $\frac{1}{2} r_0 = r_0 e^{-\lambda h}$
 $\frac{1}{2} r_0 = r_0 e^{-h/\tau}$ $\frac{1}{2} = e^{-h/\tau}$
 $-\ln 2 = -h/\tau$. Thus, $\tau = h \ln 2$

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(c) In Exercise 7, we stated that the half-life of Carbon 14 is 5230 ...

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