

## Chapter Test B Gases

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of Matter PART I In the space provided, write the letter of the term  
or phrase that best completes each statement or best answers each  
question. \_\_\_\_\_ 1. According to the kinetic-molecular theory,  
particles of an ideal gas a. attract each other but do not collide. b.

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the sum of the partial pressures of the component gases. 2. Chapter Test B Gases Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1.

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\_\_\_\_\_ 1. If the temperature of a gas remains constant, then the

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pressure of the gas will increase if the Assessment Chapter Test B

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PART I In the space provided, write the letter of the term or phrase  
that best completes each statement or best answers each question.

\_\_\_\_\_ 1. If the temperature of a gas remains constant, then the  
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Aubr3y\_Davis. Terms in this set (10) at constant volume and  
temperature the total pressure exerted by a mixture of gases is equal  
to the sum of partial pressures of the component gases.

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(b) An English unit of pressure sometimes used in engineering is  
pounds per square inch (lb/in. <sup>2</sup>), or psi: 1 atm = 14.7 lb/in. . If a  
pressure is reported as 91.5 psi, express the measurement in  
atmospheres. Answer: (a) 99.3 kPa, (b) 6.22 atm Practice Exercise 1) A  
gas at a pressure of 10.0 Pa exerts a force of \_\_\_\_\_ N on an area of

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A) 55

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number of alleles. Thus, the reproductive

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